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## X-Ray Structural Study of the Novel Tautomeric Orgaijophosphorus Compounds

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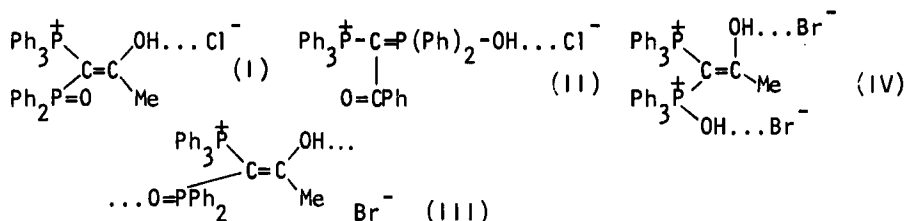
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## X-RAY STRUCTURAL STUDY OF THE NOVEL TAUTOMERIC ORGANOPHOSPHORUS COMPOUNDS

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A number of acylsubstituted (diphenylphosphoryl)methyl-triphenylphosphonium salts I-IV, which are keto-enolic organophosphorus compounds with the new type of the phosphoryl-hydroxylylide tautomerism, were studied by X-ray diffraction method.



A vast variety of tautomeric forms, occurring in crystals and stabilized by the strong H-bonds of different types, was found. Analysis of main geometric features of the studied compounds allowed to elucidate the character of electronic delocalization in the observed tautomeric forms. It was found that in I  $\text{Ph}_2\text{PO}$ -group does not essentially participate in the resonance stabilization of the cation. On the contrary, in crystal III, wherein the strong  $\text{C}-\text{OH}\cdots\text{O}=\text{P}$  H-bonds link the cations into infinite chains, the electronic system of the  $\text{Ph}_2\text{PO}$  group takes part in the stabilization of the enolic form of the cation. The geometric parameters of hydroxylylide cation also indicate the significant electronic delocalization in its central fragment. The unusual dication IV is stabilized due to the strong H-bonds of the  $\text{P}-\text{OH}$  and  $\text{C}-\text{OH}$  groups with the  $\text{Br}^-$ -anions.